



- [54] CERTAIN ALKOXY SUBSTITUTED TAXANES AND PHARMACEUTICAL COMPOSITIONS CONTAINING THEM
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- [21] Appl. No.: 863,198
- [22] Filed: Apr. 3, 1992

Related U.S. Application Data

- [63] Continuation-in-part of Ser. No. 763,805, Sep. 23, 1991, abandoned.
- [51] Int. Cl.⁵ C07D 417/08; C07D 305/14
- [52] U.S. Cl. 544/60; 544/147; 544/162; 544/379; 546/196; 546/212; 546/214; 548/525; 548/527; 549/510; 549/511
- [58] Field of Search 544/60, 147, 162, 379; 546/196, 212, 214; 548/525, 527; 549/510, 511

[56] References Cited

U.S. PATENT DOCUMENTS

4,814,470	3/1989	Colin et al.	514/449
4,857,653	8/1989	Colin et al.	549/511
4,924,011	5/1990	Denis et al.	849/510
4,924,012	5/1990	Colin et al.	849/510
4,942,184	7/1990	Haugwitz et al.	514/449
5,015,744	5/1991	Holton	549/510

FOREIGN PATENT DOCUMENTS

253738	7/1987	European Pat. Off.
253739	7/1987	European Pat. Off.
336840	4/1989	European Pat. Off.
336841	4/1989	European Pat. Off.

OTHER PUBLICATIONS

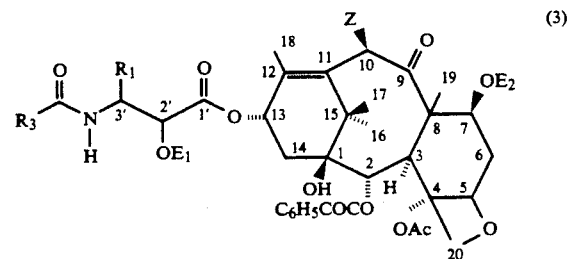
- Denis and Greene, "A Highly Efficient, Practical Approach to Natural Taxol", J. Am. Chem. Soc. 1988, 110, 5917-5919.
- Holton et al., "A Synthesis of Taxusin", J. Am. Chem. Soc., 1988, 110, pp. 6558-6560.

- Holton, "Synthesis of the Taxane Ring System", J. Am. Chem. Soc., 1984, 106, pp. 5731-5732.
- Mukerjee et al., " β -Lactams: Retrospect and Prospect", Tetrahedron vol. 34, Report No. 52, pp. 1731-1767 (1978).
- Wani et al., "Plant Antitumor Agents. VI. The Isolation and Structure of Taxol, a Novel Antileukemic and Antitumor Agent from *Taxus Brevifolia*", J. Am. Chem. Soc. 93:9, May 5, 1971, pp. 2325-2327.

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[57] ABSTRACT

A taxane derivative of the formula



wherein

R₁ is phenyl or substituted phenyl,

R₃ is CH₃O—, or CH₃CH₂O—,

Z is —OT₁,

T₁ is hydrogen, hydroxyl protecting group, or —COT₂.

T₂ is H, C₁–C₆ alkyl, C₁–C₆ alkenyl, C₁–C₆ alkynyl or monocyclic aryl,

Ac is acetyl, and

E₁ and E₂ are independently selected from hydrogen and functional groups which increase the water solubility of the taxane derivative are useful as antitumor agents.

18 Claims, No Drawings

CERTAIN ALKOXY SUBSTITUTED TAXANES AND PHARMACEUTICAL COMPOSITIONS CONTAINING THEM

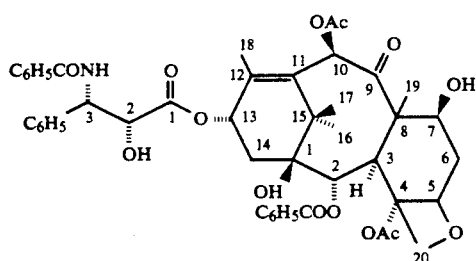
REFERENCE TO RELATED APPLICATIONS

This application is a continuation-in-part application of U.S. Ser. No. 07/763,805, filed Sep. 23, 1991 now abandoned.

BACKGROUND OF THE INVENTION

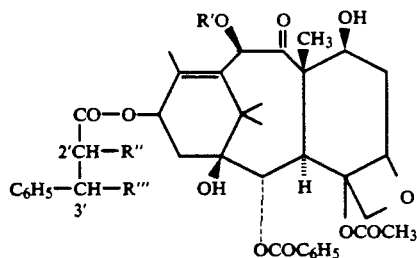
The present invention is directed to novel taxanes which have utility as antileukemia and antitumor agents.

The taxane family of terpenes, of which taxol is a member, has attracted considerable interest in both the biological and chemical arts. Taxol is a promising cancer chemotherapeutic agent with a broad spectrum of antileukemic and tumor-inhibiting activity. Taxol has a 2'R, 3'S configuration and the following structural formula:



wherein Ac is acetyl. Because of this promising activity, taxol is currently undergoing clinical trials in both France and the United States.

Colin et al. reported in U.S. Pat. No. 4,814,470 that taxol derivatives having structural formula (2) below, have an activity significantly greater than that of taxol (1).



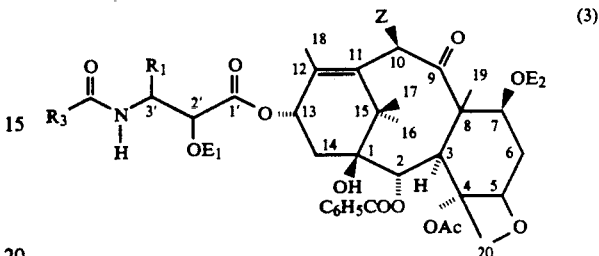
R' represents hydrogen or acetyl and one of R'' and R''' represents hydroxy and the other represents tert-butoxycarbonylamino and their stereoisomeric forms, and mixtures thereof. The compound of formula (2) in which R'' is hydroxy, R''' is tert-butoxycarbonylamino having the 2'R, 3'S configuration is commonly referred to as taxotere.

Although taxol and taxotere are promising chemotherapeutic agents, they are not universally effective. Accordingly, a need remains for additional chemotherapeutic agents.

SUMMARY OF THE INVENTION

Among the objects of the present invention, therefore, is the provision of novel taxane derivatives which are valuable antileukemia and antitumor agents.

Briefly, therefore, the present invention is directed to taxane derivatives of the formula:



wherein

R₁ is phenyl or substituted phenyl,

R₃ is CH₃O—, or CH₃CH₂O—,

Z is —OT₁,

T₁ is hydrogen, hydroxyl protecting group, or —COT₂,

T₂ is H, C₁–C₆ alkyl, C₂–C₆ alkenyl, C₂–C₆ alkynyl or monocyclic aryl,

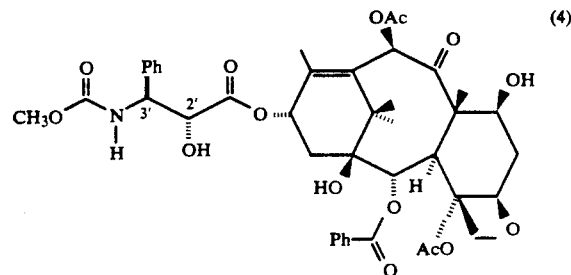
Ac is acetyl, and

E₁ and E₂ are independently selected from hydrogen, hydroxy protecting groups and functional groups which increase the water solubility of the taxane derivative.

Other objects and features of this invention will be in part apparent and in part pointed out hereinafter.

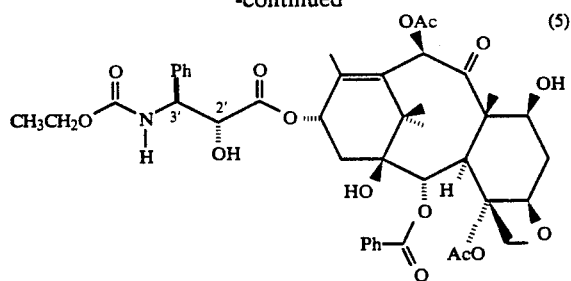
DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

In accordance with the present invention, it has been discovered that compounds having structural formula (3), in general, and structural formulas (4) and (5), in particular show remarkable properties, in vitro, and are valuable antileukemia and antitumor agents. Their biological activity has been determined in vitro, using tubulin assays according to the method of Parness et al., *J. Cell Biology*, 91: 479–487 (1981) and human cancer cell lines, and is comparable to that exhibited by taxol and taxotere.

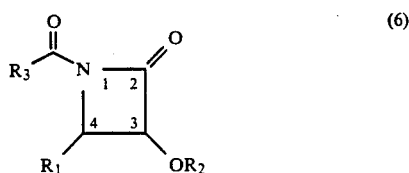


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-continued



Taxanes having formulas (4) and (5) which have the 2'R, 3'S configuration may be obtained by reacting a β -lactam with metal alkoxides having the taxane tetracyclic nucleus and a C-13 metallic oxide substituent to form compounds having a β -amido ester substituent at C-13. The β -lactams have the following structural formula:



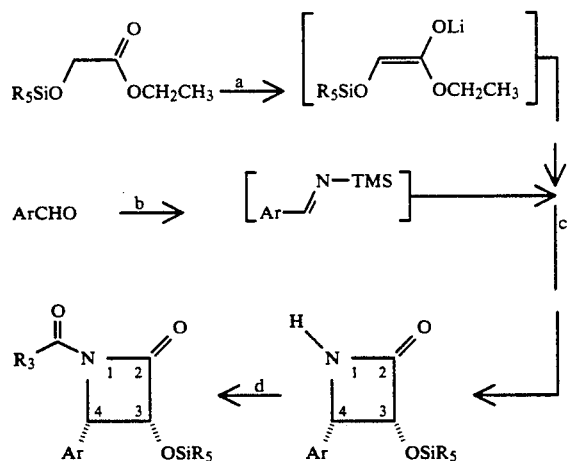
wherein

R₁ is phenyl or substituted phenyl,

R₂ is a hydroxy protecting group,

R₃ is CH₃O— or CH₃CH₂O—.

β -lactams (6) can be prepared from readily available starting materials, as is illustrated by the following reaction scheme:



reagents

(a) LDA, THF, -78°C . to -50°C .;

(b) LHMDS, THF, -78°C . to 0°C .;

(c) THF, -78°C . to 25°C ., (2h); and

(d) triethylamine and an alkyl chloroformate (R₃=alkoxycarbonyl).

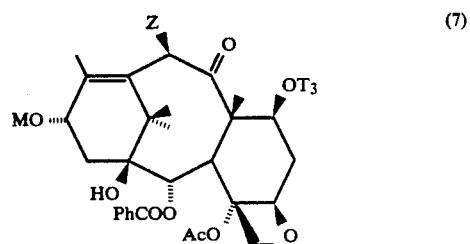
The 3-hydroxyl protecting group shown in the above reaction scheme is $-\text{SiR}_5$ wherein R₅ is trialkyl or triaryl such as triethyl. The 3-hydroxyl may be protected with other standard protecting groups such as 1-ethoxyethyl, or 2,2,2-trichloroethoxymethyl. Additional hydroxy protecting groups and the synthesis thereof may

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be found in "Protective groups in Organic Synthesis" by T. W. Greene, John Wiley & Sons, 1981.

The racemic β -lactams may be resolved into the pure enantiomers prior to protection by recrystallization of the corresponding 2-methoxy-2-(trifluoromethyl) phenylacetic esters. However, the reaction described hereinbelow in which the β -amido ester side chain is attached has the advantage of being highly diastereo-selective, thus permitting the use of a racemic mixture of side chain precursor.

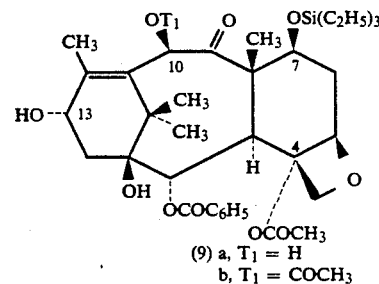
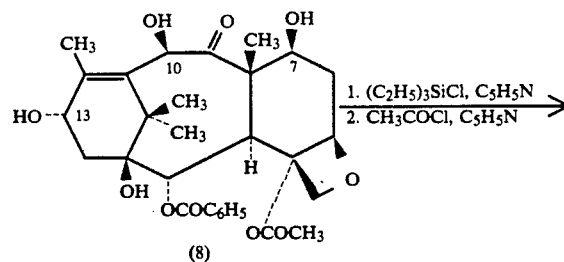
The metal alkoxides having the taxane tetracyclic nucleus and a C-13 metallic oxide substituent have the following structural formula:



wherein Z is $-\text{OT}_1$; T₁ is hydrogen, hydroxyl protecting group, or $-\text{COT}_2$; T₂ is H, C₁-C₆ alkyl, C₂-C₆ alkenyl, C₂-C₆ alkynyl or monocyclic aryl; T₃ is hydrogen or a hydroxy protecting group; and M is a metal, preferably selected from the group comprising Group IA, Group IIA and transition metals, most preferably, Li, Mg, Na, K or Ti.

Preferably, the metal alkoxides are prepared by reacting an alcohol having the taxane tetracyclic nucleus and a C-13 hydroxyl group with an organometallic compound in a suitable solvent. Most preferably, the alcohol is a protected baccatin III, in particular, 7-O-triethylsilyl baccatin III (which can be obtained as described by Greene, et al. in *JACS* 110: 5917 (1988) or by other routes) or 7,10-bis-O-triethylsilyl baccatin III.

As reported in Greene et al., 10-deacetyl baccatin III is converted to 7-O-triethylsilyl-10-deacetyl baccatin III according to the following reaction scheme:

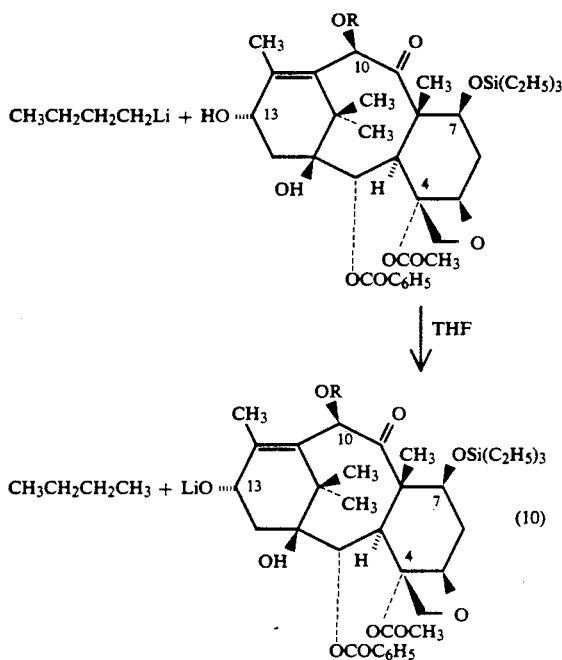


Under what is reported to be carefully optimized conditions, 10-deacetyl baccatin III is reacted with 20 equiva-

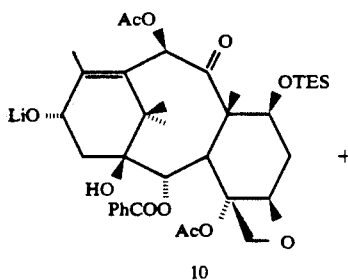
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lents of $(C_2H_5)_3SiCl$ at $23^\circ C$. under an argon atmosphere for 20 hours in the presence of 50 ml of pyridine/mmol of 10-deacetyl baccatin III to provide 7-triethylsilyl-10-deacetyl baccatin III (9a) as a reaction product in 84–86% yield after purification. The reaction product may then optionally be acetylated with 5 equivalents of CH_3COCl and 25 mL of pyridine/mmol of 9a at $0^\circ C$. under an argon atmosphere for 48 hours to provide 86% yield of 7-O-triethylsilyl baccatin III (9b). Greene, et al. in JACS 110, 5917 at 5918 (1988).

The 7-O-triethylsilyl baccatin III (9b) is reacted with an organometallic compound such as n-butyllithium in a solvent such as tetrahydrofuran (THF), to form the metal alkoxide 13-O-lithium-7-O-triethylsilyl baccatin III (10) as shown in the following reaction scheme:

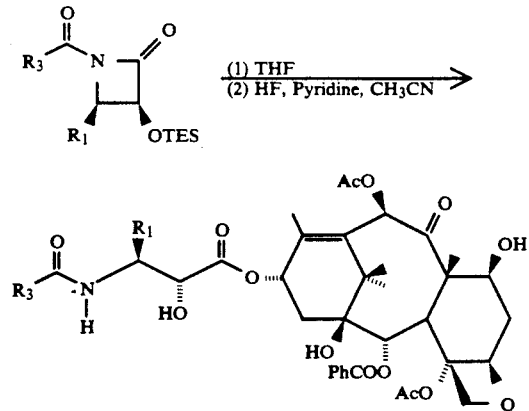


As shown in the following reaction scheme, 13-O-lithium-7-O-triethylsilyl baccatin III (10) reacts with β -lactam (6) in which R_2 is triethyl silyl to provide an intermediate in which the C-7 and C-2' hydroxyl groups are protected with a triethylsilyl group. The triethylsilyl groups are then hydrolyzed under mild conditions so as not to disturb the ester linkage or the taxane substituents.



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-continued



wherein

R_1 is phenyl or substituted phenyl,

R_3 is CH_3O- , or CH_3CH_2O- .

Both the conversion of the alcohol to the metal alkoxide and the ultimate synthesis of the taxane derivative can take place in the same reaction vessel. Preferably, the β -lactam is added to the reaction vessel after formation therein of the metal alkoxide.

The present invention also provides pharmaceutical compositions containing a compound of formula (3), in general, and the compounds of formulas (4) and (5) in particular, in combination with one or more pharmaceutically acceptable, inert or physiologically active, diluents or adjuvants.

These compositions may be presented in any form appropriate for the administration route envisaged. The parental route, and especially the intravenous route, is the preferential route for administration.

The compositions according to the invention for parenteral administration may be aqueous or nonaqueous sterile solutions, suspensions or emulsions. Propylene glycol, vegetable oils, especially olive oil, and injectable organic esters, e.g. ethyl oleate, may be used as the solvent or the vehicle. These compositions may also contain adjuvants, especially wetting agents, emulsifiers or dispersants. The sterilization may be carried out in several ways, e.g. using a bacteriological filter, by incorporating sterilizing agents into the composition, by irradiation or by heating. They may also be in the form of sterile solid compositions which may be dissolved or dispersed in sterile water or any other injectable sterile medium.

The products of general formula (3) are more particularly used in the treatment of acute leukemias and solid tumors, at daily doses which are generally between 1 and 2 mg/kg by the intravenous (perfusion) route for an adult.

The water solubility of compounds of formula (3) may be improved by modification of the C2' and/or C7 substituents to incorporate appropriate functional groups, E_1 and E_2 . For increased water solubility, E_1 and E_2 may independently be hydrogen and $-COGCOR^1$ wherein

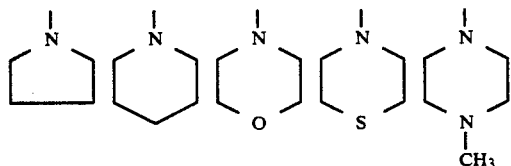
G is ethylene, propylene, $CH=CH$, 1,2-cyclohexane, or 1,2-phenylene,

$R^1=OH$ base, NR^2R^3 , OR^3 , SR^3 , $OCH_2CONR^4R^5$, OH

$R^2=$ hydrogen, methyl

$R^3=(CH_2)_nNR^6R^7$; $(CH_2)_nN^+R^6R^7R^8X^-$
 $n=1$ to 3

R⁴=hydrogen, lower alkyl containing 1 to 4 carbons
 R⁵=hydrogen, lower alkyl containing 1 to 4 carbons,
 benzyl, hydroxyethyl, CH₂CO₂H, dimethylamino-
 ethyl
 R⁶R⁷=lower alkyl containing 1 or 2 carbons, benzyl
 or R⁶ and
 R⁷ together with the nitrogen atom of NR⁶R⁷ form
 the following rings



R⁸=lower alkyl containing 1 or 2 carbons, benzyl
 X[⊖]=halide

base = NH₃, (HOC₂H₄)₃N, N(CH₃)₃, 20

CH₃N(C₂H₄OH)₂,

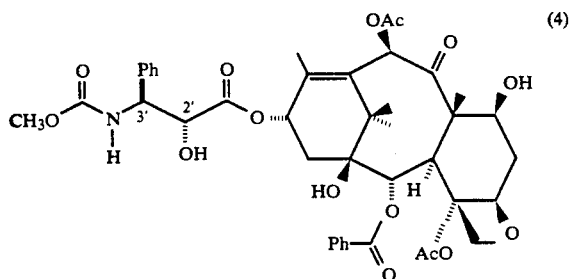
NH₂(CH₂)₆NH₂, N-methylglucamine, NaOH, KOH.

The preparation of compounds in which X₁ or X₂ is
 —COGCOR¹ is set forth in Haugwitz U.S. Pat. No.
 4,942,184 which is incorporated herein by reference.

The following examples illustrate the invention.

EXAMPLE 1

Preparation of N-debenzoyl-N-methoxycarbonyl
 taxol. 30



To a Solution of 7-triethylsilyl baccatin III (100 mg,
 0.143 mmol) in 1 mL of THF at -45° C. was added 45
 dropwise 0.087 mL of a 1.63 M solution of nBuLi in
 hexane. After 0.5 h at -45° C., a solution of cis-1-
 methoxycarbonyl -3-triethylsilyloxy-4-phenylazetid-
 2-one (252 mg, 0.715 mmol) in 1 mL of THF was added
 dropwise to the mixture. The solution was warmed to 0°
 C. and kept at that temperature for 1 h before 1 mL of a
 10% solution of AcOH in THF was added. The mixture
 was partitioned between saturated aqueous NaHCO₃
 and 60/40 ethyl acetate/hexane. Evaporation of the
 organic layer gave a residue which was purified by 55
 filtration through silica gel to give 129 mg of a mixture
 containing (2'R,3'S)-2',7-(bis)triethylsilyl-N-debenzoyl-
 N-methoxycarbonyl taxol and a small amount of the
 (2'S,3'R) isomer.

To a solution of 129 mg (0.112 mmol) of the mixture 60
 obtained from the previous reaction in 6 mL of aceton-
 itrile and 0.3 mL of pyridine at 0° C. was added 0.9 mL
 of 48% aqueous HF. The mixture was stirred at 0° C.
 for 3 h, then at 25° C. for 13 h, and partitioned between
 saturated aqueous sodium bicarbonate and ethyl acet-
 ate. Evaporation of the ethyl acetate solution gave 108
 mg of material which was purified by flash chromatog-
 raphy to give 90 mg (76%) of N-debenzoyl-N-methox-

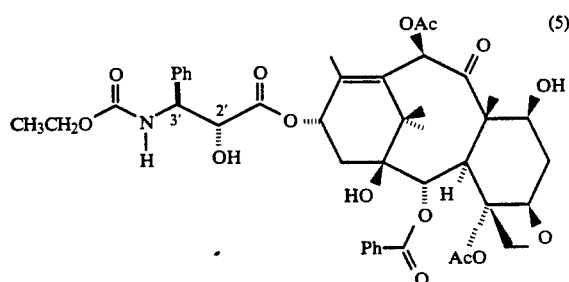
ycarbonyl taxol, which was recrystallized from me-
 thanol/water.

m.p. 169°-171° C.; [α]_D²⁵_{Na}-63.7° (c 1.1, CHCl₃).

¹H NMR (CDCl₃, 300 MHz) δ 8.12 (d, J=7.1 Hz,
 2H, benzoate ortho), 7.65-7.3 (m, 8H, aromatic), 6.28
 (m, 2H, H10, H13), 5.66 (d, J=7.1 Hz, 1H, H2β), 5.58
 (d, J=9.4 Hz, 1H, H3'), 5.30 (d, J=9.4 Hz, 1H, NH),
 4.93 (dd, J=9.3, 1.7 Hz, 1H, H5), 4.65 (br s, 1H, H2'),
 4.40 (m, 1H, H7), 4.29 (d, J=8.8 Hz, 1H, H20α), 4.17 (d,
 10 J=8.8 Hz, 1H, H20β), 3.79 (d, J=7.1 Hz, 1H, H3), 3.58
 (s, 3H, COOMe), 3.34 (d, J=5 Hz, 1H, 2'OH), 2.54 (m,
 1H, H6α), 2.47 (d, J=3.8 Hz 1H, 20H), 2.36 (s, 3H,
 4Ac), 2.24 (s, 3H, 10Ac), 2.22 (m, 2H, H14α, H14β),
 1.88 (m, 1H, H6α), 1.82 (br s, 3H, Me18), 1.73 (s, 1H,
 15 10H), 1.68 (s, 3H, Me19), 1.27 (s, 3H, Me17), 1.14 (s, 3H,
 Me16).

EXAMPLE 2

Preparation of N-debenzoyl-N-ethoxycarbonyl taxol.



To a solution of 7-triethylsilyl baccatin III (155 mg,
 0.221 mmol) in 2 mL of THF at -45° C. was added 35
 dropwise 0.136 mL of a 1.63 M solution of nBuLi in
 hexane. After 0.5 h at -45° C., a solution of cis-1-ethox-
 ycarbonyl-3-triethylsilyloxy-4-phenylazetid-2-one
 (386 mg, 1.11 mmol) in 2 mL of THF was added drop-
 wise to the mixture. The solution was warmed to 0° C.
 and kept at that temperature for 1 h before 1 mL of a
 10% solution of AcOH in THF was added. The mixture
 was partitioned between saturated aqueous NaHCO₃
 and 60/40 ethyl acetate/hexane. Evaporation of the
 organic layer gave a residue which was purified by 40
 filtration through silica gel to give 252 mg of a mixture
 containing (2'R,3'S)-2',7-(bis)triethylsilyl-N-debenzoyl-
 N-ethoxycarbonyl taxol and a small amount of the
 (2'S,3'R) isomer.

To a solution of 252 mg (0.112 mmol) of the mixture
 obtained from the previous reaction in 12 mL of aceto-
 nitrile and 0.6 mL of pyridine at 0° C. was added 1.8 mL
 of 48% aqueous HF. The mixture was stirred at 0° C.
 for 3 h, then at 25° C. for 13 h, and partitioned between
 saturated aqueous sodium bicarbonate and ethyl acet-
 ate. Evaporation of the ethyl acetate solution gave 216
 mg of material which was purified by flash chromatog-
 raphy to give 155 mg (85%) of N-debenzoyl-N-ethox-
 ycarbonyl taxol, which was recrystallized from me-
 thanol/water.

m.p. 161.5°-162.5° C.; [α]_D²⁵_{Na}-62.2° (c 0.51, CHCl₃).

¹H NMR (CDCl₃, 300 MHz) δ 8.12 (d, J=7.7 Hz,
 2H, benzoate ortho), 7.65-7.3 (m, 8H, aromatic), 6.28
 (m, 1H, H10) 6.27 (m, 1H, H13), 5.67 (d, J=7.1 Hz, 1H,
 H2β), 5.53 (d, J=9.3 Hz, 1H, H3'), 5.29 (d, J=9.3 Hz,
 1H, NH), 4.94 (dd, J=9.3, 2.2 Hz, 1H, H5), 4.64 (dd,
 J=5.0, 2.8 Hz, 1H, H2'), 4.41 (m, 1H, H7), 4.29 (d,
 J=8.5 Hz, 1H, H20α), 4.17 (d, J=8.5 Hz, 1H, H20β),

4.01 (q, $J=7.1$ Hz, 2H, $\text{COOCH}_2\text{CH}_3$), 3.79 (d, $J=7.1$ Hz, 1H, H3), 3.45 (d, $J=5$ Hz, 1H, 2'OH), 2.54 (m, 1H, H6 α), 2.47 (d, $J=3.9$ Hz 1H, 7OH), 2.36 (s, 3H, 4Ac), 2.24 (s, 3H, 10Ac), 2.22 (m, 2H, H14 α , H14 β), 1.87 (m, 1H, H6 α), 1.83 (br s, 3H, Me18), 1.77 (s, 1H, 1OH), 1.68 (s, 3H, Me19), 1.27 (s, 3H, Me17), 1.15 (s, 3H, Me16), 1.14 (t, $J=7.1$ Hz, 2H, $\text{COOCH}_2\text{CH}_3$).

EXAMPLE 3

Tubulin binding assays were performed using compounds (4) and (5) substantially as set forth in Parness et al., *J. Cell Biology* 91: 479-487 (1981) and compared to taxol and taxotere. The results are presented in Table 1.

TABLE 1

Compound Name/Formula	Tubulin Assay	
	Init. Peak	Rel. Rate
4	64	
5	44	78
Taxol	100	98
Taxotere	100	—

EXAMPLE 4

IC₅₀ data were obtained in vitro on a human cancer cell line (HCT 116) which is available from the National Cancer Institute, and a multidrug resistant cell line (HCT/VM46), which is resistant to a variety of hydrophobic agents, including taxol. Cytotoxicity was assessed in HCT116 and HCT VM46 human colon carcinoma cells by XTT (2,3-bis(2-methoxy-4-nitro-5-sulphophenyl)-5-[(phenylamino)carbonyl]-2H-tetrazolium hydroxide assay (Scudiero et al, "Evaluation of a soluble tetrazolium/formazan assay for cell growth and drug sensitivity in culture using human and other tumor cell lines", *Cancer Res.* 48:4827-4833, 1988). Cells were plated at 4000 cells/well in 96 well microtiter plates and 24 hours later drugs were added and serial diluted. The cells were incubated at 37° C. for 72 hours at which time the tetrazolium dye, XTT, was added. A dehydrogenase enzyme in live cells reduces the XTT to a form that absorbs light at 450 nm which can be quantitated spectrophotometrically. The greater the absorbance the greater the number of live cells. The results are expressed as an IC₅₀ which is the drug concentration required to inhibit cell proliferation (i.e. absorbance at 450 nm) to 50% of that of untreated control cells. The results are presented in Table 2. Lower numbers indicate greater activity.

TABLE 2

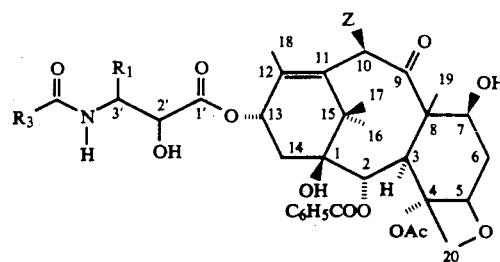
Compound Name/Formula	IC ₅₀	
	HCT 116	HCT VM46
4	0.093	n.d.
5	0.007	0.206
Taxol	0.004	0.536
Taxotere	0.007	0.246

In view of the above, it will be seen that the several objects of the invention are achieved.

As various changes could be made in the above compositions without departing from the scope of the invention, it is intended that all matter contained in the above description be interpreted as illustrative and not in a limiting sense.

What I claim is:

1. A taxane derivative of the formula



wherein

R₁ is phenyl,
R₃ is CH₃O— or CH₃CH₂O—,
Z is —OCOT₂,
T₂ is H, C₁–C₆ alkyl, C₂–C₆ alkenyl, C₂–C₆ alkynyl or monocyclic aryl, and
Ac is acetyl.

2. The taxane derivative of claim 1 wherein R₁ is phenyl and R₃ is CH₃O—.

3. The taxane derivative of claim 1 wherein, R₃ is CH₃CH₂O—, Z is —OCOCH₃, and the taxane derivative has the 2'R, 3'S configuration.

4. The taxane derivative of claim 1 wherein R₃ is CH₃CH₂O—.

5. The taxane derivative of claim 2 wherein Z is —OCOCH₃.

6. The taxane derivative of claim 1 wherein R₃ is CH₃CH₂O—, Z is —OCOCH₃, and the taxane derivative has the 2'R, 3'S configuration.

7. A pharmaceutical composition which contains the taxane derivative of claim 1 and one or more pharmacologically acceptable, inert or physiologically active diluents or adjuvants.

8. The composition of claim 7 wherein R₃ is CH₃O—.

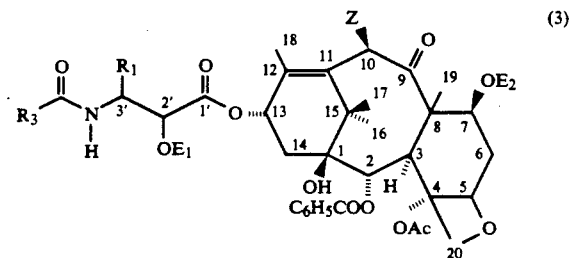
9. The composition of claim 7 wherein Z is —OCOCH₃.

10. The composition of claim 7 wherein R₃ is CH₃O—, Z is —OCOCH₃, and the taxane derivative has the 2'R, 3'S configuration.

11. The composition of claim 7 wherein R₃ is CH₃CH₂O—.

12. The composition of claim 7 wherein R₃ is CH₃CH₂O—, Z is —OCOCH₃, and the taxane derivative has the 2'R, 3'S configuration.

13. A taxane derivative of the formula



wherein

R₁ is phenyl,
R₃ is CH₃O—, or CH₃CH₂O—,
Z is OT₁,
T₁ is hydrogen, hydroxyl protecting group or —COT₂,

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T₂ is H, C₁-C₆ alkyl, C₂-C₆ alkenyl, C₂-C₆ alkynyl or monocyclic aryl,

Ac is acetyl, and

E₁ and E₂ are independently selected from functional groups which increase the water solubility of the taxane derivative, the functional groups having the formula



wherein

R is ethylene, propylene, $-\text{CH}=\text{CH}-$, 1,2-cyclohexane, or 1,2-phenylene,

R¹=OH base, NR²R³, OR³, SR³, OCH₂CONR⁴R⁵ or OH,

R²=hydrogen or methyl,

R³=(CH₂)_nNR⁶R⁷ or (CH₂)_nN[⊕]R⁶R⁷R⁸X[⊖],

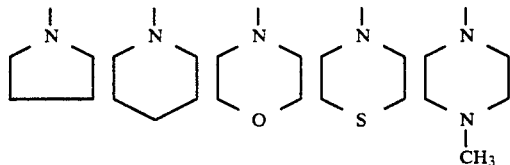
n=1 to 3,

R⁴=hydrogen, lower alkyl containing 1 to 4 carbons, or

R⁵=hydrogen, lower alkyl containing 1 to 4 carbons, benzyl, hydroxyethyl, CH₂CO₂H, dimethylaminoethyl, or

R⁶R⁷=lower alkyl containing 1 or 2 carbons, benzyl or R⁶ and

R⁷ together with the nitrogen atom of NR⁶R⁷ form the following rings



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R⁸=lower alkyl containing 1 or 2 carbons, or benzyl, X[⊖]=halide, and

base=NH₃, (HOC₂H₄)₃N, N(CH₃)₃, CH₃N(C₂H₄OH)₂, NH₂(CH₂)₆NH₂, N-methylglucamine, CaOH, or KOH.

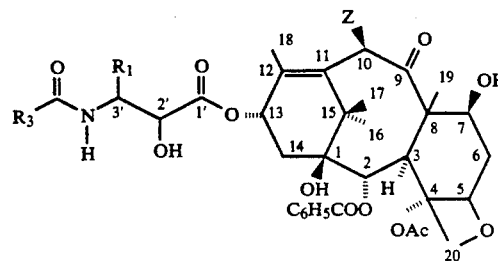
14. The taxane derivative of claim 13 wherein R₃ is CH₃O—.

15. The taxane derivative of claim 13 wherein R₃ is CH₃O— and Z is —OCOCH₃.

16. The taxane derivative of claim 13 wherein R₃ is CH₃CH₂O—.

17. The taxane derivative of claim 13 wherein R₃ is CH₃CH₂O— and Z is —OCOCH₃.

18. A taxane derivative of the formula



wherein

R₁ is phenyl,

R₃ is CH₃O—,

Z is —OT₁,

T₁ is hydrogen, hydroxyl protecting group or —COT₂,

T₂ is H, C₁-C₆ alkyl, C₂-C₆ alkenyl, C₂-C₆ alkynyl or monocyclic aryl, and

Ac is acetyl.

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UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

PATENT NO. : 5,243,045

DATED : September 7, 1993

INVENTOR(S) : Robert A. Holton et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Column 12, claim 13, line 5, "CaOH, or KOH" should read ---NaOH, or KOH---.

Signed and Sealed this
Fourteenth Day of June, 1994

Attest:



BRUCE LEHMAN

Attesting Officer

Commissioner of Patents and Trademarks